

## Modern Chemistry Section Assessment Answers

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### Modern Chemistry Section Assessment Answers

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Explanation: . Since sodium hydroxide is a strong base, it will dissociate completely in water. This means that the concentration of the base will be equal to the concentration of hydroxide ions after the reaction runs to completion.

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Explanation: . The dissociation of calcium hydroxide in aqueous solution is: The of calcium hydroxide is related to the dissolved concentrations of its counterions: and are produced in a molar ratio of 1:2; for each molecule of calcium hydroxide that dissolves:. Given a value of , the molar solubilities of each counterion may be determined by setting .

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